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Building thick spinel iron oxide layer onto the hexaferrite core nanoparticles using multiple co-precipitation of iron ions

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Introduction

- Composite nanoparticles: different functional materials combined in the single nanoparticle.
- \diamond Coupling between the different materials \rightarrow new chemical and physical properties.
- ✤ Bi-magnetic materials consist of two different magnetic materials.
- \diamond Majority of bi-magnetic nanoparticles posses core/shell structure \rightarrow strong coupling effect due to large contact area between the materials
- Thickness of the shell can influence on the magnetic properties of the composite nanoparticles



SYNTHESIS OF THE CORE/SHELL NANOPARTICLES

- \bullet high temperatures (> 200 °C).
- \diamond toxic and expensive reactants.

ALTERNATIVE

- Low temperature synthesis of the magnetic spinel iron oxide (maghemite γ - Fe_2O_3) shell/layer deposited onto the hexaferrite core nanoparticles.
- Method based on the co-precipitation of Fe^{3+}/Fe^{2+} ions and heterogeneous nucleation of the product onto the core nanoparticles [1,2].

Maximum layer thickness: 2 nm



- ✤ Investigation of the synthesis procedure aiming to increase the maghemite layer coated onto the hexaferrite core nanoparticles.
- ★ METHOD: two-step co-precipitation process of the Fe³⁺/Fe²⁺ ions in colloidal suspension of the core nanoparticles [2].

PROBLEMS

Agglomeration of the composite nanoparticles in the second step.

Decreasing of the surfce area available for the growth of the maghemite layer Agglomeration could promote formation of the homogeneously-nucleated maghemite nanoparticles.



- is possible to increase the thickness of the maghemite layer.
- Further optimization of the synthesis in needed to obtain the homogeneous product, containing only the composite nanoparticles

[2] Primc, D., B. Belec, and D. Makovec; Synthesis of composite nanoparticles using co-precipitation of a

magnetic iron-oxide shell onto core nanoparticles. J. Nanopart. Res. (18:64): p. 1-13 (2016).

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